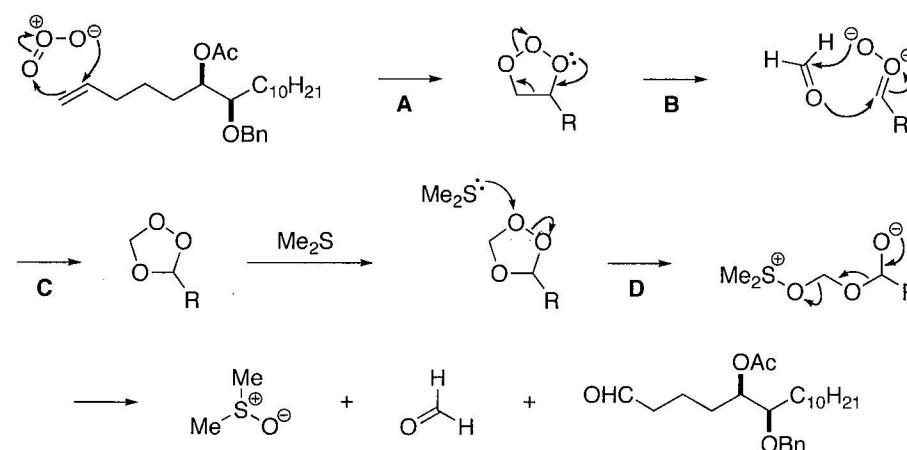


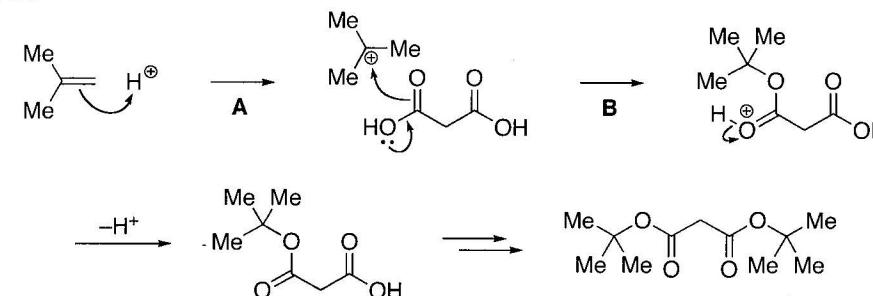
A: Hydroboration through a four-membered transition state. **B:** Attack of a hydroperoxide anion to the borane to form an ate complex. **C:** Migration of an alkyl group. **D:** Hydrolysis of the borate.

A029



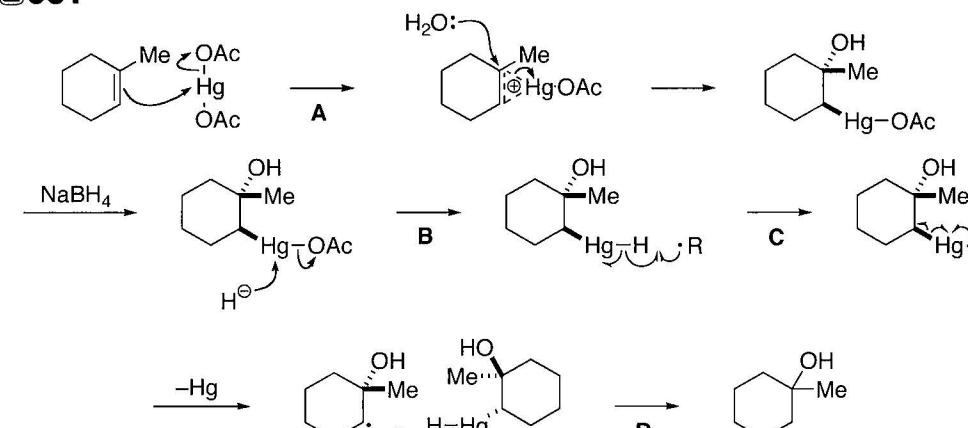
A: 1,3-Dipolar cycloaddition of ozone to the olefin. **B:** Heterolytic cleavage of the initial ozonide. **C:** Recombination of the resulting 1,3-dipole and the aldehyde to form an ozonide. **D:** Reductive cleavage of the O-O bond of the ozonide with Me_2S .

A030



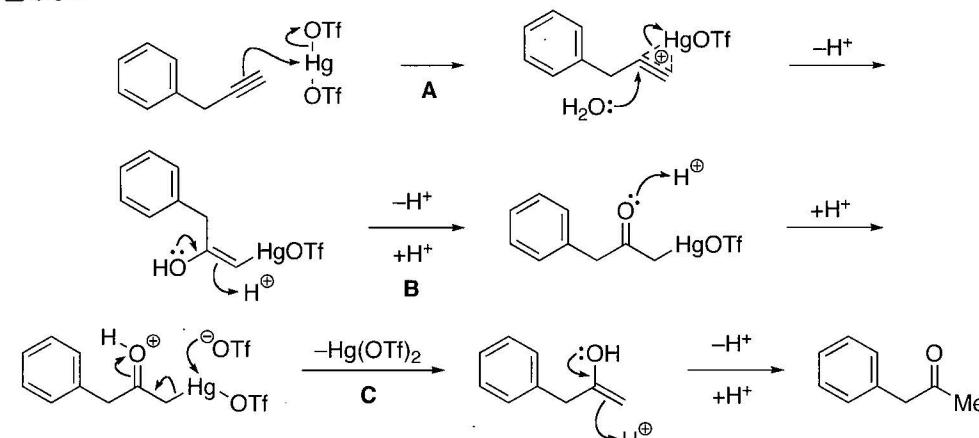
McCloskey, A. L.; Fonken, G. S.; Kluiber, R. W.; Johnson, W. S. *Org. Synth., Coll. Vol. IV* 1963, 261.
A: Protonation of isobutylene to form a stable tertiary carbocation. **B:** Attack of a carboxylic acid to the carbocation.

A031

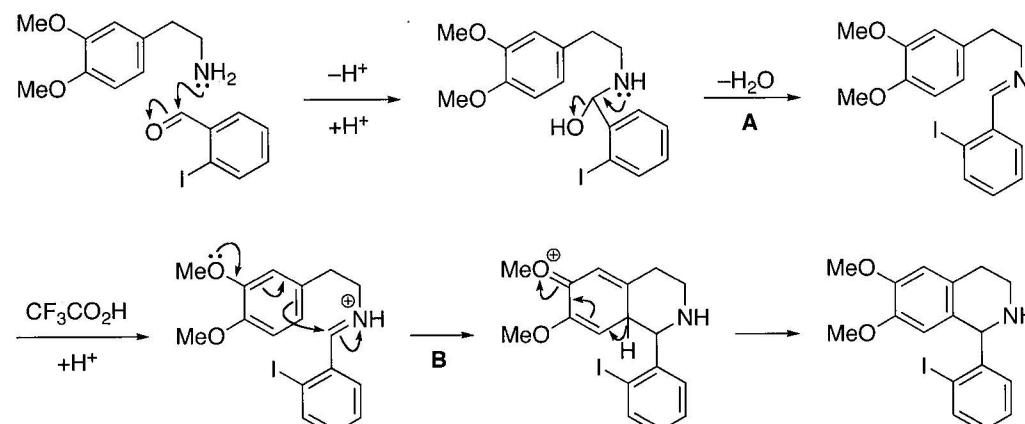


A: Oxymercuration of the olefin. **B:** Reduction with NaBH_4 to form a Hg-H bond. **C:** Cleavage of the Hg-H bond followed by extrusion of Hg to form a secondary carbon radical. **D:** Abstraction of a hydrogen atom.

A032

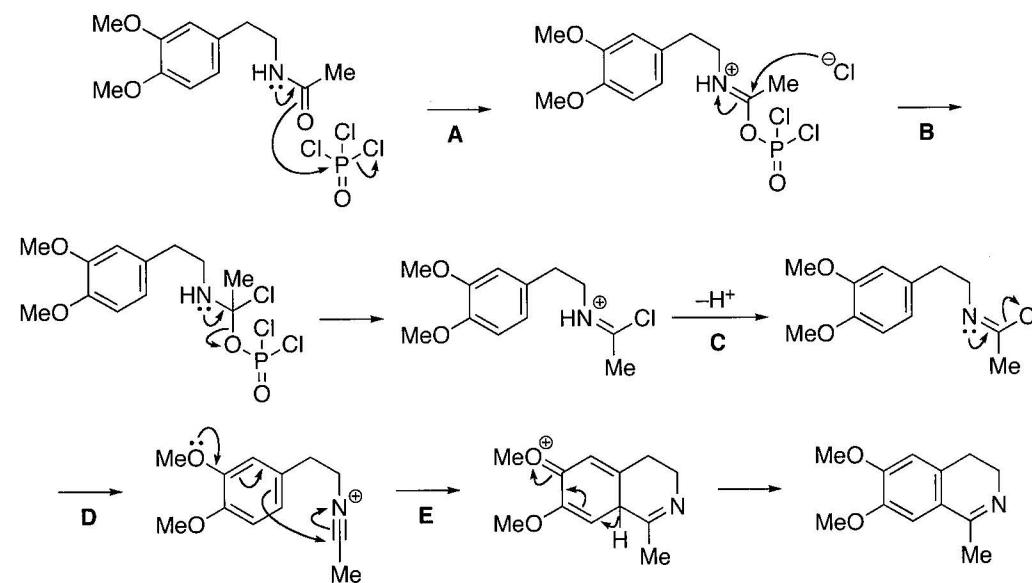


A: Oxymercuration of the alkyne. **B:** Tautomerization of the enol. **C:** Demercuration to regenerate $\text{Hg}(\text{OTf})_2$.

A033

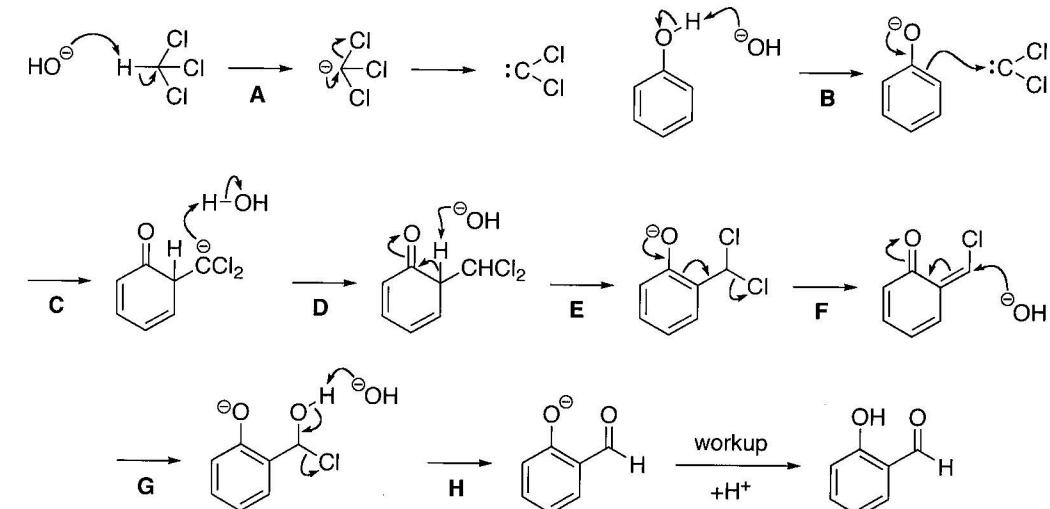
Whaley, W. M.; Govindachari, T. R. *Org. React.* **1951**, *6*, 151.

Pictet-Spengler reaction. **A:** Formation of an imine. **B:** Addition of an electron-rich aromatic ring to the iminium ion followed by aromatization.

A034

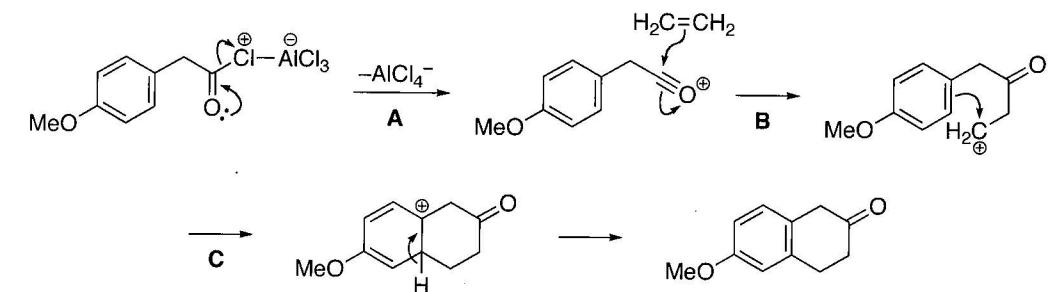
Brossi, A.; Dolan, L. A.; Teitel, S. *Org. Synth., Coll. Vol. VI* **1988**, 1.

Bischler-Napieralski reaction. **A:** Attack of the oxygen atom of the amide to POCl_3 . **B:** Addition of chloride ion followed by elimination of dichlorophosphate ion. **C:** Deprotonation. **D:** Elimination of chloride ion to form a nitrilium ion. **E:** Attack of an electron-rich aromatic ring to the nitrilium ion.

A035

Wynberg, H.; Meijer, E. W. *Org. React.* **1982**, *28*, 1.

Reimer-Tiemann reaction. **A:** Deprotonation of CHCl_3 followed by α -elimination to form dichlorocarbene ($\text{pK}_a \text{ CHCl}_3 = 13.6, \text{H}_2\text{O} = 15.7$). **B:** Formation of phenoxide ion ($\text{pK}_a \text{ PhOH} = 10$). **C:** Attack of the phenoxide ion to dichlorocarbene. **D:** Protonation. **E:** Aromatization. **F:** Elimination of chloride ion helped by the oxygen lone pair of the phenoxide ion. **G:** Conjugate addition of hydroxide ion. **H:** Elimination of chloride ion.

A036

or